

## Total synthesis of (±)-homopumiliotoxin 223G

Leonardo Silva Santos and Ronaldo Aloise Pilli\*

Instituto de Química, UNICAMP, PO Box 6154, 13083-970 Campinas, SP, Brazil Received 20 February 2001; revised 17 April 2001; accepted 27 July 2001

**Abstract**—The total synthesis of  $(\pm)$ -homopumiliotoxin 223G was achieved in six steps and 14% yield through the addition of 5-methyl-2-triisopropylsilyoxyfuran (3) to the *N*-acyliminium ion derived from *N*-Cbz 2-methoxypiperidine featuring bicyclic lactam **9b** as the key intermediate. © 2001 Elsevier Science Ltd. All rights reserved.

Amphibian skin has provided a wide range of biologically active alkaloids, many of which have unique profiles of pharmacological activity and therapeutic potential. Among them pumiliotoxins, allo- and homopumiliotoxins were shown to have myotonic and cardiotonic activity. It appears likely that these alkaloids are taken up into the skin from the diet, which for such amphibians consists mainly of small arthropods. The compounds are stored in granular skin glands and are released as defense against predators. In this communication, we report a short approach to the synthesis of homopumiliotoxin alkaloids (Fig. 1). Recently, the total synthesis of (–)-homopumiliotoxin 223G was achieved by Kibayashi et al.4

The synthetic route envisioned for homopumiliotoxin 223G features the preparation of bicyclic lactam 9, the key intermediate in our approach, through the addition of silyloxyfuran 3 to N-acyliminium ion 2.

As described in our previous communication,<sup>5</sup> the vinylogous Mannich addition<sup>6</sup> of silyloxyfuran 3<sup>7,8</sup> to the *N*-acyliminium derived from 2a,b provided *erythro*-4a,b as the major diastereoisomer when the reaction was

carried out in  $CH_2Cl_2$  and catalytic TMSOTf was used to generate the corresponding N-acyliminium ion. Different solvents and Lewis acids were sought in order to improve the yield of *threo-5a*, which displays the relative configuration required for  $(\pm)$ -1 (Scheme 1 and Table 1).

As previously noticed by Morimoto in the study of the addition of silyloxyfurans to the five-membered N-acyliminium ring, diastereoselectivity was not significantly affected neither by the nature of the Lewis acid nor by the solvent (erythro-4a:threo-5a=1.1:1–2:1), except when a binary solvent system (1:1 CH<sub>2</sub>Cl<sub>2</sub>:THF, v/v) was used. In fact, in the presence of TMSOTf the preference for the erythro isomer 4a increased (Table 1, entry 10). The use of 2b ( $R_1$ =Boc) did not improve the yield of threo-5b but provided significant amounts of regioisomer 6b (stereochemistry not determined), particularly when  $BF_3$ · $OEt_2$  was used as the Lewis acid.

The relative configurations of **4a** and **5a** were established after their conversion into the corresponding bicyclic lactam **9a** and **9b**, <sup>10</sup> followed by NOE experi-

Figure 1. Representative homopumiliotoxin alkaloids.

0040-4039/01/\$ - see front matter © 2001 Elsevier Science Ltd. All rights reserved. PII: \$0040-4039(01)01389-2

<sup>\*</sup> Corresponding author. E-mail: pilli@iqm.unicamp.br

Scheme 1. Vinylogous Mannich reaction of 3 and 2a/2b and conversion to lactams 9a and 9b.

Table 1. Addition of silyloxyfuran 3 to carbamates 2a and 2b

| Entry | R   | Solvent                             | Lewis acid                        | 4:5:6 <sup>a</sup> | Yield (%)b |
|-------|-----|-------------------------------------|-----------------------------------|--------------------|------------|
|       | Cbz | CH <sub>2</sub> Cl <sub>2</sub>     | TMSOTf                            | 2:1:0              | 85         |
| 2     | Cbz | CH <sub>2</sub> Cl <sub>2</sub>     | TiCl <sub>4</sub>                 | 1.2:1:0            | 70         |
| }     | Cbz | CH <sub>2</sub> Cl <sub>2</sub>     | BF <sub>3</sub> ·OEt <sub>2</sub> | 1.2:1:0            | 50         |
| ļ     | Cbz | Et <sub>2</sub> O                   | TMSOTf                            | 1.1:1:0            | 65         |
|       | Cbz | Et <sub>2</sub> O                   | $TiCl_4$                          | 1.2:1:0            | 58         |
|       | Cbz | Et <sub>2</sub> O                   | $BF_3 \cdot OEt_2$                | 1.1:1:0            | 50         |
|       | Cbz | THF                                 | TMSOTf                            | 1.3:1:0            | 55         |
|       | Cbz | THF                                 | TiCl <sub>4</sub>                 | _                  | _          |
|       | Cbz | THF                                 | BF <sub>3</sub> ·OEt <sub>2</sub> | 1.1:1:0            | 42         |
| 0     | Cbz | THF/CH <sub>2</sub> Cl <sub>2</sub> | TMSOTf                            | 6:1:0              | 80         |
| 1     | Cbz | THF/CH <sub>2</sub> Cl <sub>2</sub> | $TiCl_4$                          | _                  | _          |
| 2     | Cbz | THF/CH <sub>2</sub> Cl <sub>2</sub> | BF <sub>3</sub> ·OEt <sub>2</sub> | 1.4:1:0            | 43         |
| 3     | Boc | THF/CH <sub>2</sub> Cl <sub>2</sub> | TMSOTf                            | 18.5:1:11          | 67         |
| 4     | Boc | THF/CH <sub>2</sub> Cl <sub>2</sub> | $TiCl_4$                          | _                  | _          |
| 5     | Boc | THF/CH <sub>2</sub> Cl <sub>2</sub> | $BF_3 \cdot OEt_2$                | 15:1:9.8           | 40         |

<sup>&</sup>lt;sup>a</sup> Diastereoselectivity determined by GC and confirmed by <sup>1</sup>H NMR analyses.

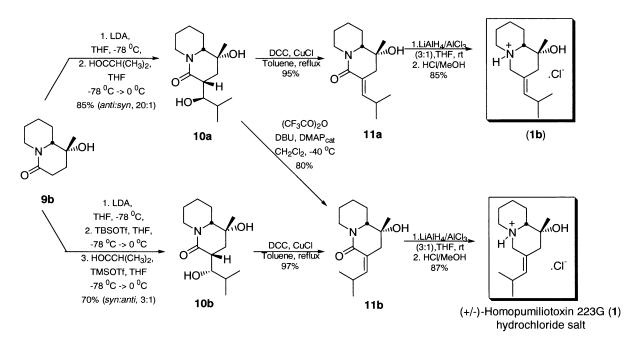
ments as described in our previous communication and next we focused on the installation of the isobutylidene moiety present in  $(\pm)$ -1.

We first explored the reaction of the preformed lithium enolate of **9b** (LDA, 2 equiv., -78°C) with isobutyraldehyde, which afforded a 20:1 mixture of two aldol adducts **10a:10b** in 85% yield. The *anti* configuration of the major isomer **10a** was determined after stereospecific *syn* elimination using DCC/CuCl, which gave **11a** in 95% yield. Accordingly, *anti* elimination of **10a** using Stork protocol afforded **11b** in 80% yield. The assignment of the double bond configuration in **11a** and **11b** was rather straightforward upon inspection of their

NMR spectra due to the characteristic deshielding of the olefinic hydrogen in **11a** ( $\delta$  6.81) when compared with the same signal in **11b** ( $\delta$  5.60) due to the anisotropy of the carbonyl group (Scheme 2).<sup>13</sup>

Alternatively, Mukaiyama aldol reaction of the *N*,*O*-silylketeneacetal derived from **9b**<sup>14</sup> and isobutyraldehyde catalyzed by TMSOTf afforded a 3:1 mixture of **10b:10a**. Stereospecific *syn* elimination using DCC/CuCl<sup>11</sup> gave **11b** in 97% yield. Once an efficient approach to bicyclic lactam **11b** was secured, the completion of the total synthesis of (±)-homopumiliotoxin 223G (1)<sup>15</sup> required AlH<sub>3</sub> reduction, <sup>16</sup> which was accomplished in 87% yield (Scheme 2). Its identity was

<sup>&</sup>lt;sup>b</sup> Yields determined after column chromatography on silica gel of the crude product.



**Scheme 2.** Total synthesis of  $(\pm)$ -homopumiliotoxin 223G.

established after comparison of its <sup>1</sup>H and <sup>13</sup>C NMR data with those reported by Kibayashi et al. <sup>4b</sup>

## Acknowledgements

We thank Fapesp and CNPq for financial support and fellowships.

## References

- Daly, J. W.; Garrafo, H. M.; Spande, T. F. In *The Alkaloids*; Cordell, G. A., Ed.; San Diego, CA: Academic Press, 1993; p. 85.
- 2. Daly, J. W. J. Nat. Prod. 1998, 61, 162.
- Jones, T. H.; Gorman, J. S. T.; Snelling, R. R.; Delabie, J. H. C.; Blum, M. S.; Garraffo, H. M.; Jain, P.; Daly, J. W.; Spande, T. F. J. Chem. Ecol. 1999, 25, 1179.
- (a) Aoyagi, S.; Hasegawa, Y.; Hirashima, S.; Kibayashi, C. *Tetrahedron Lett.* 1998, 39, 2149; (b) Kibayashi, C.; Aoyagi, S.; Wang, T. C.; Saito, K.; Daly, J. W.; Spande, T. F. J. Nat. Prod. 2000, 63, 1157.
- (a) de Oliveira, M. C. F.; Santos, L. S.; Pilli, R. A. Tetrahedron Lett. 2001, 42, 6995; (b) Santos, L. S.; Pilli, R. A. 22nd IUPAC International Symposium on the Chemistry of Natural Products, OSA-12, São Carlos, SP, Brazil (2000).
- 6. Arend, M.; Westermann, B.; Risch, N. Angew. Chem., Int. Ed. Engl. 1998, 37, 1045.
- 7. Nasman, H.; Pensar, K. G. Synthesis 1985, 786.
- 8. Martin, S. F.; Bur, S. K. Tetrahedron 1999, 55, 8905.

- Morimoto, Y.; Nishida, K.; Hayashi, Y.; Shirahama, H. Tetrahedron Lett. 1993, 34, 5773.
- 10. Analytical data. Compound 5a: <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz, 293 K):  $\delta$  1.41 (s, 3H), 1.30–1.50 (m, br, 1H), 1.50-1.70 (m, br, 3H), 1.75-1.85 (m, br, 1H), 1.90-2.05 (m, br, 1H),  $\{[2.65-2.75 \text{ (m, br)} \text{ and } 2.82 \text{ (t, } J=12.6 \text{ Hz)}],$ 1H}, {[3.91 (d, J=12.6 Hz) and 4.0–4.1 (m, br)], 1H}, 4.30 (s, br, 1H), 4.98 (d, J=12.0 Hz, 1H), 5.03 (d, J=12.0 Hz, 1H), 5.74 (d, J=5.5 Hz, 1H), 7.2–7.5 (m, 6H);  ${}^{13}$ C NMR (CDCl<sub>3</sub>, 125.7 MHz, 293 K):  $\delta$  19.4, 22.2, 23.6, 24.3, 41.1, 54.9, 67.5, 93.4, 119.4, 127.9, 128.3, 128.7, 136.8, 156.3, 160.8, 173.1. Compound **9b**: <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz, 293 K): δ 1.34 (s, 3H), 1.40-1.50 (m, 3H), 1.52–1.70 (m, 2H), 1.76 (ddd, J=5.6, 10.7 and 16.6 Hz, 1H), 1.82–1.88 (m, 3H), 2.30–2.45 (m, 2H), 2.62 (ddd, J=5.6, 10.7 and 16.1 Hz, 1H), 3.07 (d, br, J=11.3 Hz, 1H), 4.79 (d, br, J=13.1 Hz, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125.7 MHz): δ 24.3, 25.0, 26.2, 27.4, 28.5, 33.3, 43.2, 65.5, 67.9, 169.1; HRMS (70 eV): 183.1255; calcd for C<sub>10</sub>H<sub>17</sub>NO<sub>2</sub>: 183.1259.
- Fox, D. N. A.; Lathbury, D.; Mahon, M. F.; Molloy, K. C.; Gallagher, T. J. Am. Chem. Soc. 1991, 113, 2652.
- Stork, G.; Shiner, C. S.; Winkler, J. D. J. Am. Chem. Soc. 1982, 104, 310.
- Crabb, T. A.; Newton, R. F.; Jackson, D. Chem. Rev. 1971, 71, 109.
- 14. Thornton, E. R.; Stormes, M. N. J. Org. Chem. 1991, 56, 2489.
- Tokuyama, T.; Nishimori, N.; Shimada, A. *Tetrahedron* 1987, 43, 643.
- Yoon, N. M.; Brown, H. C. J. Am. Chem. Soc. 1968, 90, 2927.